

THE SNAITH SCHOOL

Science Revision Tools

Content!



A massive amount of content across Biology, Chemistry & Physics

 Students need to ensure they know what content is in each paper

 Students need to revise the content and then practice applying it to exam questions

Mock Priorities: Paper I

Highlighted are Y9 taught
HT means higher tier
T means triple only

Biology	Chemistry	Physics
B I Eukaryotes and prokaryotes Cell structure Microscopes required practical (RP) Differentiation Stem cells Cell Cycle Diffusion, Osmosis, Active Transport RP Osmosis Culturing microorganisms (T)	CI Atoms, elements, compounds, mixtures Balancing equations Development of the model of the atom Atomic structure Electronic structure Periodic table development Groups 0, & 7 Transition metals (T)	PI Energy stores and systems Kinetic energy, EPE, GPE Specific heat capacity RP Specific heat capacity Power Dissipation of energy Efficiency Energy resources RP 2: Insulation (T)
B2 Organisation Digestive System, Enzymes and Digestive Enzymes RP Food Tests RP Amylase Heart, Blood and Blood vessels Health: cancer, CHD. Plants: tissues, organs, leaf structure, Transpiration	C2 Ionic Bonding Ionic compounds Covalent bonding Giant covalent compounds including polymers, diamond and graphite Metallic Bonding States of matter and state symbols Metals and alloys Manoparticles (T)	P2 Circuit symbols Ohms Law Charge RP Resistance of a wire RP IV – and graphs Series & Parallel DC & AC Mains electricity and energy transfers National grid Static electricity (T) Electric fields (T)
B3 Bacterial, viral, fungal and protist diseases. Human defence Vaccination Antibiotics and painkillers Development of drugs Monoclonal antibodies (T) Plant diseases and defences (T)	C3 Conservation of mass Relative formula mass Calculating moles (HT) Avogadro's Constant (HT) Limiting Reactants (HT) Concentration of solutions % yield (T) Atom economy (T) Concentration in mol/dm³ (HT) (T)	P3 Density of materials RP Density Changes of state Internal energy Specific latent heat Particle motion in gases Pressure in gases (T) Increasing pressure (T) (HT)
B4 Photosynthesis: equation, factors affecting rate RP Light Intensity on pondweed Uses of glucose Aerobic respiration Anaerobic Respiration Metabolism	C4 Metal oxides Reactivity series REDOX reactions Strong and weak acids (HT) Acids and metals Neutralisation pH scale RP Making Salts Electrolysis RP Titration (T)	P4 Atomic Structure Mass number Development of atom Radioactive decay Nuclear equations Half life Radioactive contamination Background radiation (T) Nuclear fission (T) Nuclear fusion (T)
	C5 Exo and endothermic RP temperature change Reaction Profiles Bond energies (HT) Fuel cells and batteries (T)	

Exam Priorities: Paper 2

HT means higher tier T means triple only

Biology	Chemistry	Physics
B5 Homeostasis Nervous system RP Reaction Times Endocrine system Blood glucose Hormones in human reproduction Contraception IVF (HT) Feedback systems (HT) The brain (T) Body temperature (T) Kidneys (T)	C6 Calculating rates of reaction RP Rate of reaction Callision theory Catalysts Reversible reactions Le Chatelier's Principle (HT)	P5 Scalar and vector quantities Types of force Weight, mass and gravity Work Done Elasticity RP Hooke's Law Distance and displacement Speed and Velocity Acceleration RP Acceleration Newton's Laws Reaction Time and Stopping Distances Momentum (HT) Moments (T) Pressure in fluids (T)
B6 Sexual and asexual Meiosis DNA and inheritance Inherited disorders Variation Evolution and evidence Genetic engineering and selective breeding Extinction Classification Advantages of sexual and asexual reproduction (T) Structure of DNA (T) Cloning (T) Theory of evolution (T) Speciation Mendel's Genetics (T)	C7 Crude oil Fractional distillation Cracking Alkenes (T) Alcohol (T) Carboxylic Acids (T) Polymerisation (T) Minio Acids (T) DNA (T)	P6 Transverse & Longitudinal Waves Wave Properties RP Ripple Tank Electromagnetic Waves: uses and application RP: Radiation Reflection of Waves (T) Sound Waves (T) Waves for detection or exploration (T) Lenses (T) Black body radiation (T)
	C8 Pure Substances Formulations Chromatography RP Chromatography Gas Tests Flame Tests (T) Metal Hydroxides (T) RP Ion Identification (T) Flame emission spectroscopy	P7 Poles of a magnet Magnetic fields Electromagnets Left hand rule (HT) Motors (HT) Loudspeaker (T) (HT) Uses of generator effect (T) (HT) Microphones (T) (HT) Transformers (T) (HT)
B7 Communities Abiotic and Biotic Adpatations Levels of organisation RP Species Distribution Biodiversity Waste management and land	C9 Earths Atmosphere Changing atmosphere Greenhouse gases Climate change Carbon footprint Pollutants	P8 – TRIPLE ONLY Solar system Life cycle of a star Orbital motion Red-shift
use Deforestation Global warming Decomposition (T) RP Decay (T) Environmental Change (T) Trophic Levels (T) Biomass (T) Food security, farming and fishing (T) Biotechnology (T)	C10 Earths Resources Potable Water RP Water Purification Waste water treatment Life cycle assessments Corrosion (T) Alloys (T) Ceramic and composites (T) Haber Process (T) NPK Fertilisers (T)	

Homework



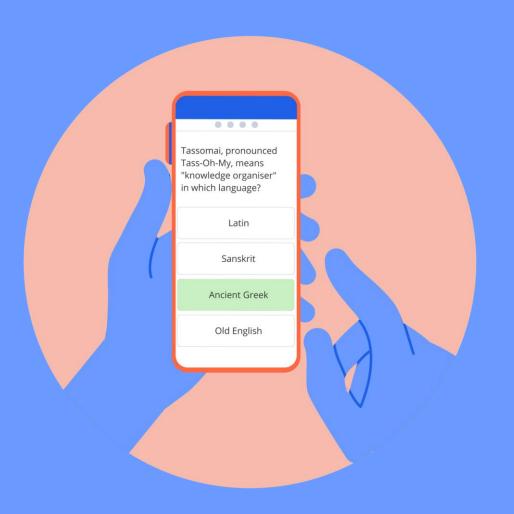
Homework is extremely important for Science as it enables students to recap prior knowledge and also to practice exam technique

There are 2 main forms of homework for Science:

- 1. Past exam question practice taken in and marked by the class teacher feedback given to improve exam technique
- 2. Tassomai

Tassomai!







How does **TASSOMAI** work?



Investigating

We start by analysing a
 subject, down to the
finest level of detail...



Creating
 turning everything a learner
needs to know into quiz questions
that teach as well as test.



3. Discovering Every question answered helps us build up a detailed knowledge profile.



Personalising
 Our intelligent algorithm
 continually adapts the content
 for each learner.



5. Supporting Sometimes we'll suggest a short tutorial video, designed to supplement learning in a particular topic.



6. Evolving Tassomai discovers more about what you understand (and what you don't) with every interaction.



7. Reinforcing
Quizzes are shown at the
optimum time and repeated
occasionally to check that
knowledge has been retained.



8. Celebrating!
Your knowledge, understanding and confidence builds until you master the subject, helping you achieve your best possible results.



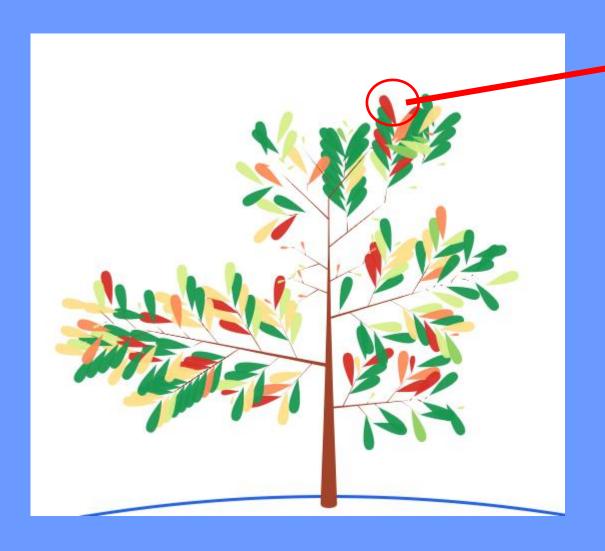
Scan the QR code to download the Tassomai app











Chemistry / The Rate & Extent of Chemical Change / Activation Energy & Catalysts

Last Seen: 16/07/25

Last 3 Attempts: $\bigotimes \bigotimes \bigotimes$







A CATALYST has the effect of ---(1)--- for a reaction. This will decrease the distance between the ---(2)--- on a REACTION PROFILE.

1: decreasing the ACTIVATION ENERGY; 2: REACTANTS and MAXIMUM energy levels

1: decreasing the ACTIVATION ENERGY; 2: REACTANTS and PRODUCTS energy levels

1: increasing the rate of COLLISIONS; 2: REACTANTS and MAXIMUM energy levels

1: increasing the rate of COLLISIONS; 2: REACTANTS and PRODUCTS energy levels











The Learning Program

What does the parent dashboard offer?



Parents can monitor their children's **Daily Goal completion** across the week (and go back by up to 3 weeks' worth of activity). They can also control their account settings, and opt into our new Parent Club! **Learn more here**.

https://www.tassomai.com/school-parents

It will also highlight the **weakest areas** for their child across their subjects, and **provide resources** to help support learning.





Command Words in GCSE Science

Command	Meaning	Example 1	Example 2
		Both graphite and graphene comprise	Both aerobic and anaerobic respiration
	Describe both the	hexagonal rings of carbon atoms that are	use glucose and release energy although
	similarities and differences	covalently bonded to three other carbon	aerobic respiration releases more. Also,
Compare	between two or more	atoms in a giant lattice structure. Graphene	aerobic respiration produces carbon
things.		is different to graphite because it is a 2D	dioxide and water whereas anaerobic
	tilligs.	structure as it's only one atom thick.	
		structure as it's only one atom thick.	respiration produces lactic acid.
	B	The state of the s	(E.g. include data from the graph in your
	Recall accurate facts about	The plum pudding model of the atom	answer). The temperature increased for
Describe* an object, eve process.	an object, event or	consists of a ball of positive charge with	1 minute before remaining constant for
	process.	negative electrons dotted throughout	two minutes. It then rose again for three
			minutes.
		The student will need to use the same	
	Set out how something	volume and concentration of acid to	
Design	will be done.	ensure a fair test. They can change the	
	will be done.	temperature using a water bath and	
		measure how much gas is produced in 20s.	
		0	Power stations that emit less carbon
			dioxide cause less global warming. Coal
		One advantage of using hydrogen fuel cells	and geothermal power stations
		is that the only emission produced is water.	contribute to global warming, whereas
	Use your own knowledge	Petrol makes carbon dioxide and nitrogen	nuclear power doesn't. Sulfur dioxide
Evaluate	and the information	oxides when burnt. Carbon dioxide is a	causes acid rain meaning coal power
(often a 6 mark	provided to come up with	greenhouse gas and nitrogen oxides	stations contribute to acid rain,
question)	arguments for and against	contribute to acid rain. One disadvantage is	whereas geothermal and nuclear power
440000000	something and then add a	hydrogen is difficult to store. Overall, I	stations don't. Radioactive waste, which
	summary.	think fuel cells are the better option,	has a long half-life and remains
	\$1000A110A110A110A1	provided we can find a suitable source of	radioactive for a long time, needs
		hydrogen.	burying. Overall, I think nuclear power is
		.,,	the best option, provided we can store
			its waste material safely.
		Potassium is more reactive than sodium	Bile helps the digestion of milk by
	State something more	because potassium has more electron	neutralising acid and emulsifying fats so
Explain*	clearly; give the reason(s)	shells resulting in a weaker attraction	that they have a greater surface area
Explain	for it.	between the nucleus and the outer	thereby allowing enzymes to work more
	TOTAL.	electron meaning it is lost more easily.	effectively.
		Glass is a much better material for milk	errectivery.
	Lles suidenes (dete		
	Use evidence (data,	bottles than plastic because each bottle	
Justify	statements, quotes) from	can be re-used 25 times instead of once	
	the question to support	and new bottles can be made using up to	
	your answer.	50% recycled material. Plastic bottles are	
		only made up of 10% recycled material.	
		Add an excess of the metal oxide to the	Measure the object's mass using a mass
		acid and stir to make sure all of the acid	balance. Fill a displacement can with
D I		has reacted. Filter the mixture to remove	water, ensuring it is levelled with the
Plan	VA/mite a manth and		can's spout. Place the object in the can
(often a 6 mark	Write a method.	the excess oxide. Heat the filtrate to form a	and collect the displaced water in a
question)		saturated solution. Transfer to a	measuring cylinder to determine its
		crystallising dish and leave for the crystals	volume. Calculate density =
		to slowly form as the water evaporates.	mass/volume
	Use evidence (data,		
	statements, quotes) from	The nail in test tube two rusted the most.	
Show	the question and/or your	You can tell this because the mass	
Snow	own knowledge to reach a	increased by the greatest amount from	
	conclusion.	8.45g to 8.91g.	
	100.000 ACC 100.000 ACC 100.000	The student sould shange the mathed to	
	Apply your knowledge and	The student could change the method to	Suggest how needing less oxygen helps
Suggest*	understanding from a	investigate the rate of reaction at 40°C by	the animal to conserve water. Less water
	similar situation to the	putting both of the reactants in a water	is lost from respiration.
	one presented.	bath.	

^{*}These are the most common command words



Physics - units



Students **will** full equation sheet, they do not need to memorise the equations.

However, if students can learn the units of each term in Physics, this will massively help them to understand which equation they need to choose to use in a question e.g.

Calculate the gravitational potential energy of a 12kg box at a height of 2.5m above the floor. gravitational field strength = 9.8 N/kg

Using flashcards



Repetition strategy - practice makes perfect!

 Simple approach - 'cue' on the front and the 'answer' on the back

Engage active recall of information

Why flashcards help you learn



 They help you to recall information - creating stronger connections for your memory

 Encourage self-reflection - embedding information into your long term memory

Make Flashcards More Powerful



Retrieve (don't look) - write or say the answer out loud before flipping the card over

Reorder (shuffle) - to add challenge

- spacing and interleaving

Repeat (at least 3 times) - to make sure you really know it and remember it!

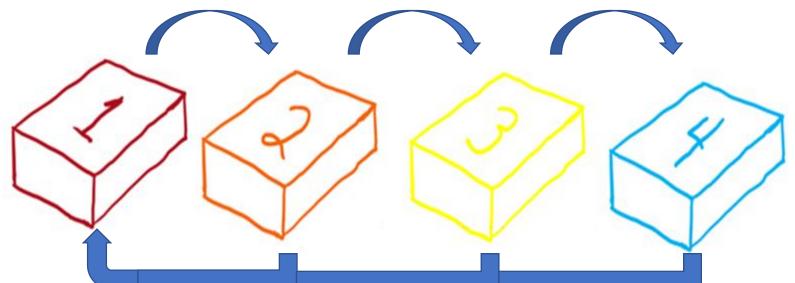
Top tips for using flashcards



Use **spaced repetition** - review on specific days of increasing intervals (day 1, day 2, day 4, day 8...) as this activates your long term memory

All the cards start off in box 1

As you review the cards, each card you answer correctly goes into box 2



As you continue to get the answers correct, move the cards to box 3 and then 4.

Any incorrect answers in any box, go back to box 1

GCSE Past Papers



- Google "AQA Past Papers"
- https://www.aqa.org.uk/find-past-papers-and-mark-schemes
- link to this on TEAMS for students

Find past papers and mark schemes Find past papers and mark schemes to help you prepare for exams. Select a subject to start your search. Subject Science Qualification GCSE Specification Combined Science: Trilogy (8464) Series All available series

A range of revision strategies



- Tassomai quizzes
- Flashcards
- GCSE Past Papers
- Exam practice workbook
- Videos GCSEPod/YouTube e.g. cognito science, kay science or free science lessons
- Posters/Mind maps linking topics together
- Revision apps



Any Questions?